

Ejercicios de nomenclatura Química Inorgánica

1. KH
2. CaH₂
3. NiH₂
4. CuH
5. HgH
6. BH₃
7. SiH₄
8. H₂S
9. HF
10. H₂Te
11. HCl (aq)
12. H₂S (aq)
13. Ni₂O₃
14. Br₂O₅
15. Cl₂O
16. PbO
17. Na₂O
18. CuO
19. SnO₂
20. SO₃
21. N₂O
22. As₂O₃
23. MgO
24. NO
25. NaCl
26. NiSe
27. BF₃
28. Cu₂S
29. HgSe
30. Fe₂S₃
31. BaCl₂
32. Ni₂S₃
33. K₂Te
34. PtF₂
35. NiP
36. BN
37. SiC
38. SF₆
39. HClO₄
40. HNO₃
41. H₃PO₃
42. HNO₂
43. H₂S₂O₅
44. HBrO
45. H₄SiO₄
46. HAsO₂
47. HPO₃
48. H₃PO₄
49. HNO₂
50. H₄P₂O₇
51. HClO₂
52. H₂S₂O₅
53. H₃AsO₃
54. H₂CrO₄
55. HIO
56. Co(OH)₃

Ejercicios de nomenclatura Química Inorgánica

- | | |
|----------------------------------|---|
| 57. $\text{Fe}(\text{OH})_3$ | 80. $(\text{Fe}(\text{H}_2\text{PO}_4)_3$ |
| 58. $\text{Ba}(\text{OH})_2$ | 81. $\text{Ni}(\text{HSO}_3)_3$ |
| 59. $\text{Cu}(\text{OH})_2$ | 82. AgHCO_3 |
| 60. $\text{Fe}(\text{OH})_3$ | 83. NH_4HSO_3 |
| 61. $\text{Pt}(\text{OH})_2$ | 84. $\text{Cu}(\text{HSiO}_3)_2$ |
| 62. $\text{Al}(\text{OH})_3$ | 85. $\text{Fe}(\text{HS})_2$ |
| 63. LiOH | 86. $\text{Cu}(\text{HSe})_2$ |
| 64. NaClO_3 | 87. $\text{Co}(\text{H}_2\text{AsO}_3)_3$ |
| 65. K_2SO_4 | 88. FeHPO_4 |
| 66. $\text{Fe}(\text{SO}_3)_3$ | 89. Au^{3+} |
| 67. $\text{Co}(\text{NO}_2)_2$ | 90. IO_3^- |
| 68. $\text{Pb}_3(\text{PO}_4)_2$ | 91. ClO^- |
| 69. CuNO_3 | 92. Se^{2-} |
| 70. $(\text{NH}_4)_2\text{SO}_3$ | 93. Cu^{2+} |
| 71. NaNO_3 | 94. Fe^{3+} |
| 72. SnS_2O_7 | 95. ClO^- |
| 73. NaClO | 96. NH_4^+ |
| 74. $\text{Fe}(\text{BrO}_3)_3$ | 97. SO_3^{2-} |
| 75. Ag_2CrO_4 | 98. S^{2-} |
| 76. FeCO_3 | 99. NO_2^- |
| 77. NH_4NO_3 | 100. CO_3^{2-} |
| 78. K_2SO_3 | 101. ClO_3^- |
| 79. CaHPO_4 | |